PATENT SPECIFICATION

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NO DRAWINGS

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COMPLETE SPECIFICATION

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Method of Dyeing Hair

We, SCHWARZKOPF VERWALTUNG G.M.B.H., a body corporate and existing under the laws of Germany, of Hohenzollernring 127—129, Hamburg-Altona, Germany, trading as HANS SCHWARZKOPF, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described

in and by the following statement:—
This invention relates to oxidation hair dyes, i.e. hair dyes which are applied to the hair and oxidised in situ to develop the dye

In our U.K. patent specification No.

15 1,026,978 there is disclosed and claimed a method of dyeing hair in which the hair is treated with a neutral or slightly alkaline composition containing 2,5-diaminepyridine. After application the dye is developed by oxidation, for example, by means of hydrogen peroxide. By this method a good red colour can be obtained which is highly resistant to moisture.

In accordance with the present invention it has been found that equally satisfactory results can be obtained using 2,5-diaminopyridines substituted in the 2-amino group. In particular the dyes according to the present invention are excellent for obtaining red and orange nuances in the hair which have a high resistance to washing, light and diffusion, are uniform in the colours obtained and are stable over a wide pH range. These dyes are therefore markedly superior to the previously used nitro compounds.

The substituted 2,5-diaminopyridines used in this invention are of the formula

H₂N N R₁

where R₁ represents hydrogen or an alkyl, cycloalkyl, hydroxyalkyl, aminoalkyl, alkoxyalkyl or aryl group;

R₂ represents an alkyl, cycloalkyl, hydroxyalkyl, aminoalkyl, dialkylaminoalkyl, aralkylalkoxyalkyl aryl or aminopyridyl group.

Typical substituents on the 2-amino nitrogen include methyl, hydroxyethyl, cyclohexyl, methoxypropyl, dimethylaminopropyl, benzyl and phenyl. These derivatives may be used either alone, combined with one another or combined with knewn benzenoid oxidation dyes or pyridine oxidation dyes.

In contrast to unsubstituted 2,5-diaminopyridine, which can be used to provide only a red tint, the use of the substituted derivatives in accordance with this invention opens up quite new nuances for genuine tinting. In general the following rule can be postulated concerning the influence of substituents in the 2-amino group of 2,5-diaminopyridine:

The colouring effect increases from orange through red to blue-violet in the following order of substituents:

Red-orange→ yellowish red-brown→ red-brown→

brownish red-violet

The preparation of the compounds used in this invention takes place by known methods and does not form part of the invention. For example, the 2-alkylamino-5-aminopyridine compounds may be formed by reacting 2,5-diaminopyridine with alkyl halides, alkyl sulphates or diazomethane or by the method of Tschitschibabin in the presence of sodium amide. 2-Dimethylamino-5-aminopyridine can be obtained in good yield by reacting 2-amino-5-nitropyridine with methyl iodide in the presence of sodium amide and then reducing the resultant compound, similar to the method of Tschitschibabin and Knunjanz.

As in the parent case, the dye compound is applied to the hair in a neutral or slightly alkaline medium, preferably an ammoniacal medium, and may be a solution, cream or paste. After application the dye compound, which is colourless, is developed by oxidation to give the desired colour. This oxidation may result simply from atmospheric oxygen in which case rather lighter shades develop, or oxidation may be effected by using a chemical oxidant, usually hydrogen peroxide, which may be applied to the hair separately or it may be mixed with the dye composition immediately before application. The latter oxidation method results in deeper fashionable hair shades.

In addition to the dye compound, the com-

positions will generally contain conventional additives such as thickeners, stabilisers or emulsifiers. In the Examples which follow such additives are not included for the sake of simplicity.

EXAMPLE 1.
A solution is made up from:

2.0 g. 2 - hydroxyethylamino - 5 - aminopyridine - hydrochloride 4.0 g. 25% ammonia and 40

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4.0 g. 25 % ammonia and 94.0 g. water.

The solution is applied to bleached hair, left to act for 30 minutes during which time atmospheric oxidation takes place. The hair is then rinsed and dried. The result is a beautiful purple shade with good resistance to wetting.

EXAMPLE 2.

70 g. of the solution from Example 1 is mixed with 30 ml 6%, hydrogen peroxide and the mixture is allowed to act on the hair. After a dyeing time of 20 minutes at room temperature the hair is rinsed. An extremely deep stable purple is obtained.

The following table summarises the dyeing performance of other derivatives of 2,5-diaminopyridine using compositions and dyeing procedures similar to those described in Examples 1 and 2.

No.	Formula		
3	H ₃ N CH ₃	2-dimethylamino-5- 5-amino-pyridine	blue-violet
4.	H ₂ N CH ₃ C ₂ H ₄ OH	2-(N-methyl-N- hydroxyethyl)amino- 5-aminopyridine	brownish red-violet
5	H ₂ N H	2-cyclohexylamino- 5-aminopyridine	yellowish red-brown
6	H ₂ N H CH ₂ ·CH ₂ ·CH ₂ ·OCH ₃	2-(\gamma-methoxypropyl)- amino-5-aminopyridine	red-brown
7	H ₂ N CH ₃ CH ₂ .CH ₂ .CH ₂ .N-CH ₃	2-(γ-dimethylamino- propyl)amino-5- aminopyridine	Bluish red-brown
8	H ₂ N N N N N N N N N N N N N N N N N N N	2-phenylamino-5- aminopyridine	brownish red-violet

EXAMPLE 9. A solution is made up from: 2.0 g. 2 - dimethylamino - 5 - aminopyridine; 1.0 g. 2 - methyl - hydroxyethylamino - 5-5 where Ri represents hydrogen or an alkyl, aminopyridine. 4.0 g. 25% ammonia cycloalkyl, hydroxyalkyl, aminoalkyl, alkoxyalkyl or aryl group, and R2 represents an alkyl, 93 g. water. cycloalkyl, hydroxyalkyl, aminoalkyl, dialkyl-After application and developing as in example aminoalkyl, aralkyl, alkoxyalkyl, aryl or amino-10 1 ne hair takes on a quiet violet shade. Depyridyl group, followed by oxidation to developed as in example 2 (with H2O2) a deep velop the dye. bordeaux red tint is obtained. 2. A method according to claim 1, in which the said pyridine derivative is 2 - dimethyl-EXAMPLE 10. amino - 5 - aminopyridine, 2 - (N - methyl-A solution is made up from: N - hydroxyethyl)amino - 5 - aminopyridine, 50 2 - cyclohexylamino - 5 - aminopyridine, 2-1.0 g. p-toluylenediamine sulphate (γ - methoxypropyl)amino - 5 - aminopyridine, 15 1.0 g. 2 - methyl - hydroxyethylamino - 5-2 - (y - dimethylaminopropyl)amino - 5aminopyridine aminopyridine or 2 - phenylamino - 5 - amino-94.0 g. water. pyridine. 4.0 g. 25% ammonia 3. A method according to claim 1, in which the composition also contains a benzenoid or The solution is applied to hair and developed pyridine compound known as an oxidation hair as in example 2. The hair is dyed a light dye. 4. A method according to any one of the chestnut shade. preceding claims, in which the composition is Example 11. an ammoniacal solution, cream or paste. A solution is made up from: 5. A method according to any one of the preceding claims, in which the dye is de-2.0 g. 2 - hydroxyethyl amino - 5 - aminoveloped by means of a chemical oxidant other 25 pyridine - hydrochloride than atmospheric oxygen. 0.1 g. 2,6-diaminopyridine 6. A method according to claim 5, in which 93.9 g. water. the oxidant is hydrogen peroxide. 4.0 g., 25% ammonia 7. A method according to claim 5 or 6, in which the oxidant is mixed with the dye com-70 The solution is used as in Example 2, to proposition immediately prior to application to the duce a very natural looking ash blonde tint. In this case the 2,6-diaminopyridine has acted 8. A method according to claim 1, substantially as described in any one of the foregoing as a modifier.

WHAT WE CLAIM IS: -

1. A method of dyeing hair in which the hair is treated with a neutral or slightly alkaline composition containing one or more substituted 2,5-diaminopyridines of the formula:

Examples.

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9. Hair when dyed by a method claimed in any one of the preceding claims.

For the Applicants, D. YOUNG & CO., Chartered Patent Agents, 9. Staple Inn, London, W.C.1.

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